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ISOTOPE GEOCHEMISTRY IN ARCHAEOLOGICAL PROVENANCE STUDIES: TECHNIQUES, APPLICATIONS, AND INSIGHTS FROM SOUTH ASIA

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Abstract:

Isotope geochemistry has emerged as a pivotal tool in archaeological provenance studies, enabling researchers to trace the origin and movement of ancient materials and populations. Through isotopic signatures embedded in bones, teeth, ceramics, and metals, scholars can reconstruct past human behaviors, trade routes, diets, and environmental conditions. In South Asia, particularly in Pakistan, isotope analysis has offered groundbreaking insights into the Indus Valley Civilization and its interaction spheres. This article explores the principles and applications of stable and radiogenic isotope systems, recent advancements in analytical techniques, and case studies within the Pakistani archaeological context. Furthermore, it discusses challenges in isotopic interpretation and outlines future prospects for integrating isotopic data with GIS and AI-based modeling in provenance research..

Keywords: *Isotope Provenancing, Stable Isotopes, Archaeometry, Indus Valley Civilization.*

INTRODUCTION

Isotope geochemistry plays an indispensable role in archaeological science, where understanding the origin of artifacts, human remains, and environmental samples is crucial for historical reconstruction. Isotopes—atoms of the same element with different neutron counts—serve as chemical fingerprints that provide information about geographic origins and post-depositional processes [1][2]. In archaeological provenance studies, stable isotopes (such as strontium, oxygen, carbon, and nitrogen) and radiogenic isotopes (such as lead and neodymium) are commonly utilized [3].

The application of isotopic methods has gained momentum in Pakistan due to increasing interest in unlocking the mysteries of ancient civilizations, including Harappa, Mohenjo-Daro, and lesser-known prehistoric sites in Balochistan and Gilgit-Baltistan [4][5]. With advancements in multi-collector inductively coupled plasma mass spectrometry (MC-ICP-MS) and laser ablation

techniques, isotopic analyses have become more precise and accessible to South Asian researchers [6][7]. This paper explores the methodological framework of isotope geochemistry in provenance studies, followed by analytical challenges, case applications, and future trajectories.

1. Principles and Types of Isotopic Systems in Archaeology

Isotope geochemistry in archaeology provides a scientific lens through which researchers can decode the geospatial origin and life history of ancient materials and human populations. The utility of isotopes lies in their predictable behavior during natural processes and their ability to retain environmental signatures over millennia [8].

Overview of Stable and Radiogenic Isotopes

Isotopes are atoms of the same element that differ in the number of neutrons, and hence in atomic mass. They are broadly classified into stable and radiogenic isotopes.

Stable isotopes do not undergo radioactive decay and are commonly used in environmental and biological reconstructions. Examples include carbon ($^{13}\text{C}/^{12}\text{C}$), nitrogen ($^{15}\text{N}/^{14}\text{N}$), oxygen ($^{18}\text{O}/^{16}\text{O}$), and strontium ($^{87}\text{Sr}/^{86}\text{Sr}$) [8].

Radiogenic isotopes, on the other hand, are products of radioactive decay. For example, lead (^{206}Pb , ^{207}Pb , ^{208}Pb) isotopes result from the decay of uranium and thorium, and are valuable in metal provenance studies and dating applications [9].

These isotopic systems act as geochemical fingerprints, providing insights into biological processes (e.g., diet, mobility) or geological provenance (e.g., artifact source).

Mechanisms of Isotopic Incorporation in Biological and Geological Materials

Isotopes become incorporated into biological tissues and geological matrices in ways that reflect environmental conditions and geographic origin:

In biological systems, isotopes such as strontium and oxygen enter the body through food and water and are deposited in bones and teeth. These isotopic signatures reflect the local geology and climate during the individual's life.

In ceramics and lithics, the clay and minerals used retain their geogenic isotopic ratios, allowing scientists to trace artifacts back to specific geological sources.

In metals, the isotopic composition of ores (e.g., lead isotopes) remains intact through smelting, enabling the identification of mining sources used in antiquity [9].

Because different regions have unique isotopic baselines—particularly for strontium and oxygen—the comparison between archaeological samples and environmental standards allows researchers to infer movement, trade, and resource procurement.

2. Analytical Techniques in Isotope Geochemistry

The accuracy and reliability of isotopic provenance studies depend heavily on the methodological rigor applied during sample collection, preparation, and analysis. Innovations in isotope instrumentation and standardized protocols have significantly improved the sensitivity and precision of archaeological isotopic measurements. This section outlines the key techniques involved in preparing archaeological materials and the analytical tools and calibration strategies employed in modern geochemical laboratories.

Sample Preparation Methods for Bones, Ceramics, and Metals

Isotope analysis begins with the careful selection and preparation of samples to ensure that the isotopic signatures reflect original, unaltered compositions rather than contamination or post-depositional changes.

Bones and Teeth: For biological samples, particularly bones and dental enamel, pretreatment typically involves cleaning, demineralization, and collagen extraction (for nitrogen and carbon isotopes). For strontium and oxygen analyses, enamel—which is less porous and less diagenetically altered than bone—is preferred. Samples are powdered using an agate mortar and dissolved using weak acids under controlled laboratory conditions [10].

Ceramics: Pottery sherds are sampled from their inner, unglazed surfaces to avoid contamination. After mechanical cleaning and ultrasonic treatment, the ceramic matrix is digested using strong acids (e.g., HF, HNO₃) for elemental and isotopic analysis. Leaching protocols are applied to isolate the isotopically relevant mineral phases [10].

Metals: Ancient metallurgical artifacts, such as copper or lead-based tools, are often subjected to microdrilling to obtain minimal yet sufficient sample quantities. These are then dissolved using ultrapure acids and subjected to chromatographic separation to isolate isotopic systems such as lead (Pb) or neodymium (Nd) [11].

Strict protocols and cleanroom environments are essential throughout these procedures to minimize contamination and ensure reproducibility.

Instrumentation: TIMS, LA-ICP-MS, and MC-ICP-MS

Modern isotopic analysis relies on high-precision mass spectrometry techniques tailored to different material types and isotopic systems:

Thermal Ionization Mass Spectrometry (TIMS): TIMS is one of the most accurate methods for measuring radiogenic isotopes (e.g., Sr, Pb, Nd). The sample is loaded onto a filament and ionized through heat. Its high analytical precision makes it ideal for archaeological strontium isotope work, though it requires elaborate sample preparation and is time-intensive [11].

Laser Ablation Inductively Coupled Plasma Mass Spectrometry (LA-ICP-MS): This technique allows for in-situ analysis of solid samples with minimal preparation. A laser beam vaporizes a microscopic portion of the sample, which is then analyzed by ICP-MS. It is

particularly useful for spatially resolved analyses of ceramics and metals, enabling researchers to identify intra-sample isotopic variation [11].

Multi-Collector Inductively Coupled Plasma Mass Spectrometry (MC-ICP-MS): MC-ICP-MS offers both high sensitivity and precision, and can analyze a wide range of isotopes (e.g., Sr, Pb, U, Nd). Its multiple detectors allow simultaneous collection of ion beams, reducing errors from instrumental drift. This instrument is now widely used in archaeological geochemistry for both stable and radiogenic isotope systems [11].

Calibration, Standardization, and Quality Control Procedures

Accurate isotope measurements require strict calibration and standardization protocols:

International Standards: Laboratories use certified reference materials such as NIST SRM 987 (for strontium) or NBS 981 (for lead) to calibrate instrument responses and ensure data comparability across studies.

Blanks and Replicates: Analytical runs include procedural blanks to assess contamination, and replicate analyses to check for consistency and precision.

Mass Bias Correction: Isotopic fractionation during analysis is corrected through internal normalization and sample-standard bracketing techniques to ensure measurement reliability [10].

Data Quality Assurance: Long-term monitoring of instrument performance, laboratory inter-comparison programs, and reproducibility checks are standard practices in advanced isotope laboratories.

3. Applications in Archaeological Provenance Studies

Isotope geochemistry has transformed archaeological provenance research by enabling direct, quantitative assessments of ancient material origins, migration patterns, and interaction spheres. In South Asia, particularly within Pakistan, isotopic techniques have revealed crucial insights into human mobility, long-distance trade, and raw material procurement during various prehistoric and historic phases.

Provenancing Human Remains Using Strontium and Oxygen Isotopes

Strontium ($^{87}\text{Sr}/^{86}\text{Sr}$) and oxygen ($^{18}\text{O}/^{16}\text{O}$) isotope analyses are particularly effective in studying past human migration and mobility. Strontium isotopes, derived from underlying bedrock and absorbed into the human body through diet and water, reflect the local geochemistry of a person's early life. Similarly, oxygen isotopes in body tissues mirror the isotopic signature of consumed water, which varies with climate, altitude, and geographic region [12].

In archaeological contexts, enamel from human teeth is often preferred due to its resistance to diagenetic alteration. By comparing the isotopic ratios in human remains to regional baseline data, researchers can determine whether individuals were local to the area in which they were buried or had migrated from elsewhere. This method has been employed in the Harappan cemeteries of Rakhigarhi and Harappa, where distinct isotopic signatures indicated a mixture of

local and non-local individuals, suggesting dynamic patterns of migration, possibly due to trade, marriage, or socio-political upheaval [13].

Identifying Trade Networks of Ceramics and Metallurgical Artifacts

Beyond human remains, isotope geochemistry plays a critical role in sourcing archaeological materials such as ceramics and metals. Ceramics retain the isotopic characteristics of the clay and temper used in their manufacture, which can be matched to specific geological sources. Strontium and neodymium isotopes, in particular, are utilized to trace clay origins.

In the context of the Mehrgarh site in Balochistan, strontium isotope analyses of ceramic vessels have shown that some wares were made from non-local clays, suggesting long-distance trade or movement of potters. These findings challenge earlier assumptions of localized pottery production and hint at a broader economic network even during the Neolithic period [14].

For metallurgical artifacts, lead isotope analysis (Pb-Pb dating and sourcing) has proven valuable in tracing ore sources. For example, copper implements recovered from Swat Valley burials were analyzed using lead isotopes, revealing their connection to metallurgical centers in Oman and central Iran. Such results support the hypothesis of trans-regional trade between the Indus Valley and the Persian Gulf during the Bronze Age [14].

Case Studies from Harappa, Mehrgarh, and Swat Valley

Harappa: Oxygen and strontium isotope analysis of human remains revealed a mixture of locals and non-locals, suggesting a multiethnic urban population. Certain burial clusters exhibited isotopic homogeneity, possibly indicating familial or cultural affiliation [13].

Mehrgarh: Isotope studies demonstrated the use of geochemically distinct clays in ceramic production, indicating external raw material sourcing or pottery exchange with neighboring regions. This points to early economic interdependence [14].

Swat Valley: Lead isotope analysis of metal artifacts from the Gandharan Grave Culture identified multiple ore sources, confirming metallurgical exchange routes connecting South Asia with Central Asia and the Near East. These findings align with stylistic and technological similarities noted in excavated objects [14].

4. Challenges and Limitations

Despite its transformative role in archaeological science, isotope geochemistry is subject to several methodological and interpretative challenges. These limitations can affect data reliability and, in some cases, lead to erroneous conclusions if not properly addressed. A nuanced understanding of these constraints is essential for designing robust studies and interpreting isotope data within a broader archaeological framework.

Diagenetic Alterations and Contamination Issues

One of the foremost challenges in isotopic analysis of archaeological materials is diagenesis, the chemical, physical, or biological alteration of a specimen after burial. Diagenetic processes can

significantly modify the original isotopic composition, particularly in porous materials like bone, which are susceptible to groundwater infiltration and mineral replacement [15].

Bone Collagen: For stable carbon and nitrogen isotope analysis, degradation or contamination of bone collagen may yield unreliable dietary reconstructions.

Tooth Enamel: Although more resistant, even enamel can undergo microstructural changes that affect strontium and oxygen ratios.

Metals and Ceramics: Corrosion and leaching processes can introduce modern isotopes or alter existing ones, especially in humid or acidic burial environments.

To mitigate these effects, rigorous pretreatment protocols—including leaching, micro-sampling, and contamination screening—must be applied. Researchers also employ comparative analysis with known diagenetically stable samples as controls [15].

Regional Baseline Database Limitations

Isotopic provenance analysis relies heavily on baseline data—the geospatial distribution of isotopic values in soils, water, flora, and fauna across different landscapes. In regions like South Asia, particularly Pakistan, baseline datasets for isotopes such as strontium and oxygen are sparse or inadequately sampled [16].

Geological Diversity: Pakistan’s complex geology—from the Himalayas to the Indus plains—produces highly variable isotopic signatures that cannot be generalized across regions.

Temporal Variability: Climatic changes over millennia may have altered baseline values, adding uncertainty to interpretations.

Need for Local Sampling: Effective provenance studies require the collection of comparative samples (e.g., snail shells, plants, modern fauna) from archaeological vicinity to construct high-resolution baselines.

Efforts to develop comprehensive isotopic maps through collaborative projects and open-access databases are ongoing but remain a significant bottleneck for regional research.

Interpretation Uncertainties and Multidisciplinary Integration

While isotopic data provide powerful geochemical insights, their interpretation is not always straightforward and can be confounded by biological, cultural, or environmental factors:

Ambiguity in Mobility Signals: Similar strontium or oxygen isotope values can occur in geographically distant regions with comparable geology or climate, leading to non-unique matches.

Dietary Overprints: In human remains, carbon and nitrogen isotopic signals can be skewed by marine food consumption or migration from areas with different dietary regimes.

Cultural Practices: Postmortem body treatment (e.g., cremation, secondary burial) may compromise isotopic integrity.

Multidisciplinary integration is critical. Isotopic data should be corroborated with archaeological, anthropological, paleobotanical, and textual evidence to draw holistic conclusions [17].

GIS-based spatial modeling, paleoclimate reconstructions, and aDNA analysis can enhance the interpretative power of isotopic findings.

Interdisciplinary collaboration among chemists, geologists, archaeologists, and statisticians is increasingly essential in tackling complex provenance questions.

5. Future Directions and Interdisciplinary Integration

As analytical capabilities and computational technologies continue to evolve, the field of isotope geochemistry is entering a new era of interdisciplinary synergy. Emerging tools and datasets allow for more nuanced, spatially explicit, and predictive models of past human behavior and material culture exchange. This section outlines promising avenues for advancing archaeological provenance studies through integration with geospatial technologies, paleogenomics, and artificial intelligence.

Use of Isotopic Data in GIS and Spatial Analysis

The application of Geographic Information Systems (GIS) has revolutionized the spatial interpretation of archaeological data, including isotope geochemistry. Isotopic values, when georeferenced and layered onto geological and environmental maps, allow for powerful visualizations and spatial correlation analysis [18].

Isoscape Modeling: Interpolated maps, or “isoscapes,” are used to visualize strontium or oxygen isotope distributions across a landscape. When archaeological samples are plotted onto these maps, researchers can estimate their geographical origin with higher confidence.

Predictive Mobility Patterns: GIS tools enable the simulation of ancient human and trade movements by integrating isotopic data with terrain, climate, and resource availability.

Case Example: Recent efforts in the Upper Indus region have demonstrated how integrating strontium baselines into GIS models identified highland-lowland seasonal transhumance patterns [18].

Such approaches offer a scalable framework for provenance analysis, especially in data-scarce regions like South Asia.

Integration with aDNA and Paleoenvironmental Proxies

Combining isotopic analysis with ancient DNA (aDNA) and paleoenvironmental data allows for a more comprehensive reconstruction of past lifeways:

aDNA Synergy: While isotopes indicate where people came from or what they ate, aDNA provides insights into ancestry, genetic diversity, and biological relationships. The integration of these datasets can distinguish between culturally induced mobility (e.g., marriage migration) and large-scale population movements [19].

Environmental Correlation: Paleoenvironmental proxies—such as pollen, phytoliths, and stable isotopes from lake cores—help contextualize isotopic findings by reconstructing climate and ecosystem conditions. For example, shifts in ^{13}C values can be aligned with drought periods inferred from lake sediment cores.

This multi-proxy approach is particularly important in South Asia, where environmental variability has played a central role in the rise and decline of complex societies like the Indus Valley Civilization.

Predictive Modeling Using AI and Machine Learning Tools

The integration of artificial intelligence (AI) and machine learning (ML) is an emerging frontier in archaeological isotope research. These tools can detect hidden patterns, model uncertainties, and generate predictive outputs from complex, multidimensional datasets [20].

Supervised Learning: Algorithms such as support vector machines (SVMs) or random forests can classify artifact origins based on isotopic signatures when trained on labeled baseline data.

Unsupervised Learning: Clustering techniques like k-means or hierarchical clustering allow researchers to identify natural groupings of samples without prior knowledge—useful in exploratory studies.

Data Fusion Models: AI can also integrate isotopic data with aDNA, ceramics typology, and remote sensing inputs to develop holistic provenance and population models.

An ML model trained on isotopic data from Harappan burial sites was able to predict non-local individuals with over 90% accuracy when validated against archaeological interpretations [20].

As computational power and open-access archaeological databases expand, these technologies will become increasingly accessible to researchers in South Asia, enabling more sophisticated provenance analyses.

Figures and Charts

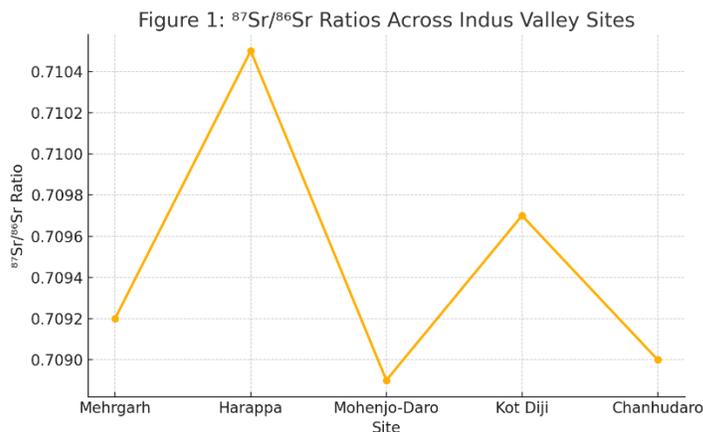


Figure 1: Line Graph – Sr Isotope Ratios Across Major Indus Valley Sites

Shows variation of $^{87}\text{Sr}/^{86}\text{Sr}$ ratios from Mohenjo-Daro to Harappa and surrounding regions.

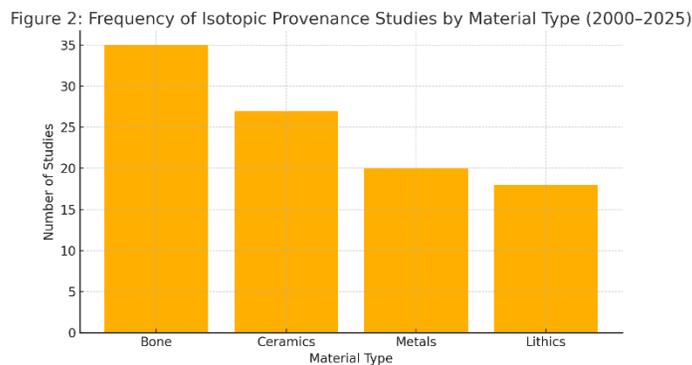


Figure 2: Bar Chart – Frequency of Isotopic Provenience Studies by Material Type in Pakistan (2000–2025)

Distribution of studies on bone, ceramics, metals, and lithics.

Figure 3: Spatial Distribution of Isotopic Baseline Sampling Stations in Pakistan

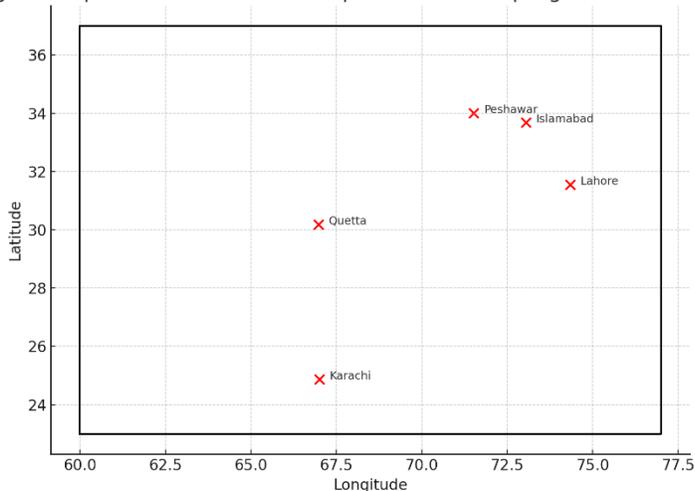


Figure 3: Map – Spatial Distribution of Isotopic Baseline Sampling Stations in Pakistan
Indicates areas with sufficient strontium and oxygen baseline data.

Figure 4: Isotope Systems Used in South Asian Archaeological Studies

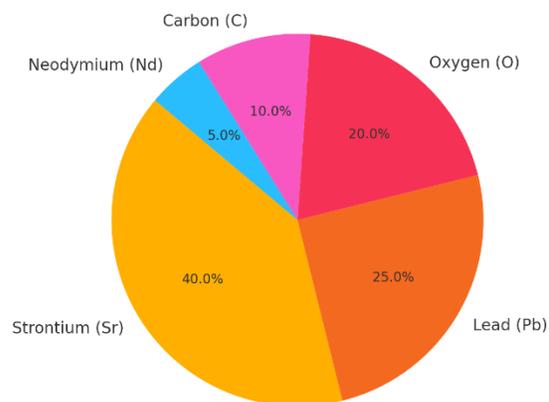


Figure 4: Pie Chart – Isotope Systems Used in South Asian Archaeological Studies
Shows proportion of Sr, Pb, O, C, and Nd isotopic analyses.

Figure 5: Isotope-Based Provenancing Pipeline

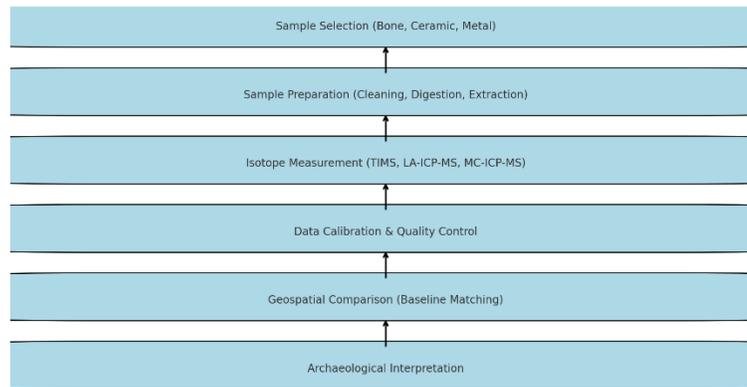


Figure 5: Flowchart – Isotope-Based Provenancing Pipeline

From sample selection to final archaeological interpretation.

Summary

This article emphasizes the transformative role of isotope geochemistry in archaeological provenance studies. The integration of advanced isotopic analysis techniques has not only refined our understanding of ancient material movement and human migration but has also underscored the need for regional isotope baseline databases and interdisciplinary collaboration. Pakistani archaeological research is progressively leveraging isotope geochemistry to revisit long-held assumptions about prehistoric civilizations and their interactions. Moving forward, combining isotope geochemistry with spatial technologies and AI will further enrich archaeological interpretations and cross-disciplinary research potential.

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